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FILE COVERS 1907 - 30 Oct 2006 VOL 145 ISS 19
FILE LAST UPDATED: 29 Oct 2006 (20061029/ED)

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L8 ANSWER 1 OF 3 HCAPLUS COPYRIGHT 2006 ACS on STN
AN 2004:430508 HCAPLUS
DN 141:9609
TI Lithium secondary battery
IN Okumura, Takefumi; Nishimura, Shin; Iwayasu,
Norio; Yokoyama, Shoichi; Yabe, Takeshi
PA Japan
SO U.S. Pat. Appl. Publ., 14 pp., Cont.-in-part of U.S. Ser. No. 623,497.
CODEN: USXXCO
DT Patent
LA English
FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI	US 2004101759	A1	20040527	US 2003-717646	20031121
	US 2004101758	A1	20040527	US 2003-623497	20030722
PRAI	JP 2002-337790	A	20021121		
	US 2003-623497	A2	20030722		

AB The object of the present invention is to provide a lithium secondary battery of high output. According to the present invention, there is provided a lithium secondary battery having a pos. electrode and a neg. electrode which reversibly intercalate and deintercalate lithium and an electrolyte containing an ion conductive material and an electrolytic salt, where the electrolyte contains an electrolytic salt and a boron-containing compound represented by the following formula $Z1(AO)mOB(O(AO)nZ2)O(AO)pZ3$ where, B is boron atom, Z1, Z2, and Z3 are the organic groups having an acryloyl group or a methacryloyl group; AO represents an oxyalkylene group of C1-6 and comprises one, or two or more of the oxyalkylene groups; and m, n and p each represents an average degree of polymerization of the oxyalkylene

group and are >0 and <4 provided that $m+n+p \geq 1$.

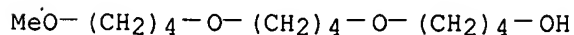
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693782-30-8P 693782-31-9P 693782-32-0P

RL: DEV (Device component use); SPN (Synthetic preparation); PREP
(Preparation); USES (Uses)
(lithium secondary battery)

RN 693782-27-3 HCAPLUS
CN Boric acid (H3BO3), 4-[4-(4-methoxybutoxy)butoxy]butyl
4-[4-[(2-methyl-1-oxo-2-propenyl)oxy]butoxy]butyl ester (9CI) (CA INDEX
NAME)

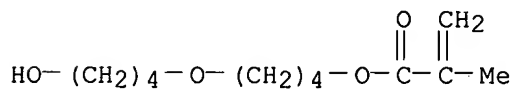
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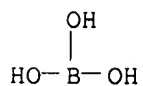
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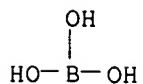
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CN Boric acid (H3BO3), 2-[2-(2-methoxyethoxy)ethoxy]ethyl
2-[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]ethyl ester (9CI) (CA INDEX
NAME)

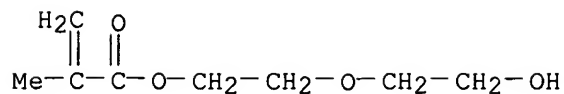
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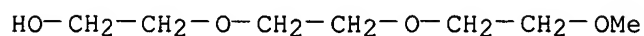
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CMF C8 H14 O4



CM 3

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CMF C7 H16 O4



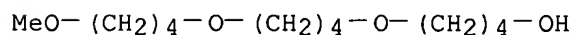
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CN 2-Propenoic acid, 2-methyl-, 4-(4-hydroxybutoxy)butyl ester, ester with boric acid (H3BO3) 4-[4-(4-methoxybutoxy)butoxy]butyl ester, homopolymer (9CI) (CA INDEX NAME)

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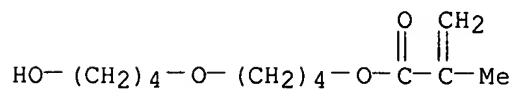
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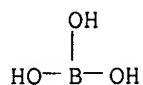
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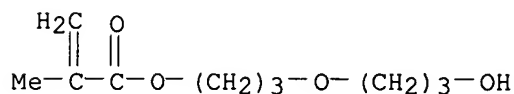
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 CN Boric acid (H3BO3), 3-[3-(3-methoxypropoxy)propoxy]propyl
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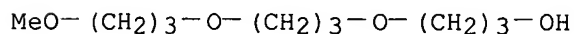
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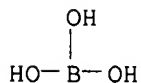
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 CMF B H3 O3



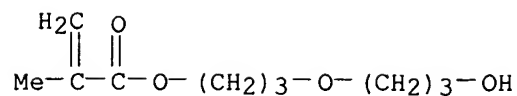
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 CN 2-Propenoic acid, 2-methyl-, 3-(3-hydroxypropoxy)propyl ester, ester with
 boric acid (H3BO3) 3-[3-(3-methoxypropoxy)propoxy]propyl ester,
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CM 1

CRN 693782-30-8
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CM 2

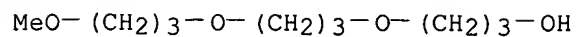
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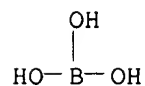
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CRN 10043-35-3

CMF B H3 O3



RN 693782-32-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(2-hydroxyethoxy)ethyl ester, ester with boric acid (H3BO3) 2-[2-(2-methoxyethoxy)ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

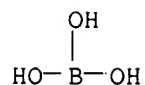
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CMF C8 H14 O4 . x C7 H16 O4 . x B H3 O3

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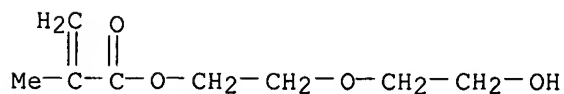
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CM 3

CRN 2351-43-1

CMF C8 H14 O4



CM 4

CRN 112-35-6
CMF C7 H16 O4HO-CH₂-CH₂-O-CH₂-CH₂-O-CH₂-CH₂-OMe

L8 ANSWER 2 OF 3 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2004:430507 HCAPLUS

DN 141:9608

TI Lithium secondary battery

IN Okumura, Takefumi; Nishimura, Shin; Iwayasu,
Norio; Yokoyama, Shoichi; Yabe, Takeshi

PA Japan

SO U.S. Pat. Appl. Publ., 14 pp.

CODEN: USXXCO

DT Patent

LA English

FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2004101758	A1	20040527	US 2003-623497	20030722
	FR 2847721	A1	20040528	FR 2003-13581	20031120
	FR 2847721	B1	20060804		
	KR 2004045326	A	20040601	KR 2003-82489	20031120
	CN 1503398	A	20040609	CN 2003-10118013	20031120
	US 2004101759	A1	20040527	US 2003-717646	20031121
	JP 2004186150	A2	20040702	JP 2003-391808	20031121
PRAI	JP 2002-337790	A	20021121		
	US 2003-623497	A2	20030722		

AB The object of the present invention is to provide a lithium secondary battery of high output. According to the present invention, there is provided a lithium secondary battery having a pos. electrode and a neg. electrode which reversibly intercalate and deintercalate lithium and an electrolyte containing an ion conductive material and an electrolytic salt, where the electrolyte contains an electrolytic salt and a boron-containing compound represented by the formula Z1(AO)mOB(O(AO)nZ2)O(AO)pZ3 or a polymer thereof (where B is a boron atom; Z1, Z2, and Z3 are organic groups having an acryloyl group or a methacryloyl group; AO represents an oxyalkylene group of C1-6 and comprises one or two or more of the oxyalkylene groups; and m, n and p each represent an average degree of polymerization of the oxyalkylene group and are 0-4).

IT 693782-27-3P 693782-28-4P 693782-29-5P

693782-30-8P 693782-31-9P 693782-32-0P

RL: DEV (Device component use); SPN (Synthetic preparation); PREP
(Preparation); USES (Uses)

(lithium secondary battery)

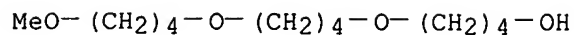
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CN Boric acid (H3BO3), 4-[4-(4-methoxybutoxy)butoxy]butyl
4-[4-[(2-methyl-1-oxo-2-propenyl)oxy]butoxy]butyl ester (9CI) (CA INDEX
NAME)

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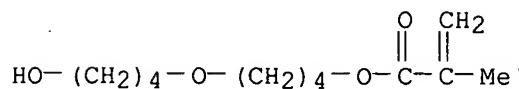
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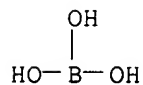
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CMF B H3 O3



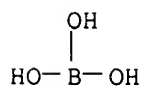
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CN Boric acid (H3BO3), 2-[2-(2-methoxyethoxy)ethoxy]ethyl
2-[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]ethyl ester (9CI) (CA INDEX
NAME)

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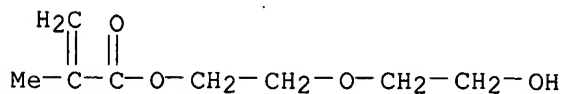
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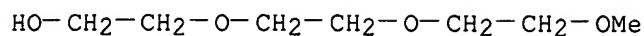
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CMF C8 H14 O4



CM 3

CRN 112-35-6
CMF C7 H16 O4



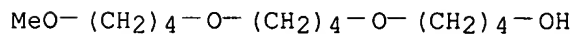
RN 693782-29-5 HCAPLUS
CN 2-Propenoic acid, 2-methyl-, 4-(4-hydroxybutoxy)butyl ester, ester with boric acid (H3BO3) 4-[4-(4-methoxybutoxy)butoxy]butyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

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CMF C13 H28 O4 . x C12 H22 O4 . x B H3 O3

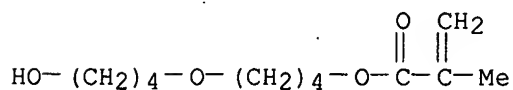
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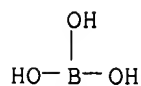
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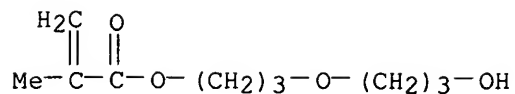
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CMF B H3 O3



RN 693782-30-8 HCAPLUS
CN Boric acid (H3BO3), 3-[3-(3-methoxypropoxy)propoxy]propyl 3-[3-[(2-methyl-1-oxo-2-propenyl)oxy]propoxy]propyl ester (9CI) (CA INDEX NAME)

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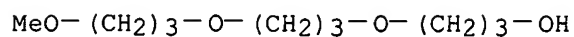
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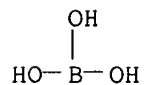
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CM 3

CRN 10043-35-3

CMF B H3 O3



RN 693782-31-9 HCAPLUS

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boric acid (H3BO3) 3-[3-(3-methoxypropoxy)propoxy]propyl ester,
homopolymer (9CI) (CA INDEX NAME)

CM 1

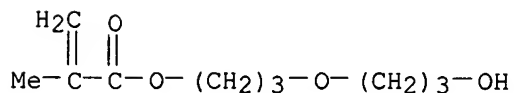
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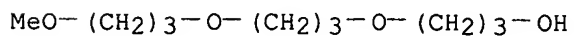
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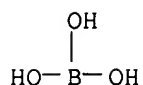
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CRN 10043-35-3

CMF B H3 O3



RN 693782-32-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(2-hydroxyethoxy)ethyl ester, ester with boric acid (H3BO3) 2-[2-(2-methoxyethoxy)ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

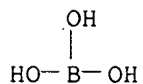
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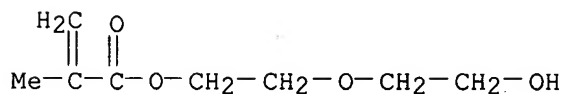
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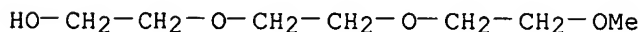
CMF C8 H14 O4



CM 4

CRN 112-35-6

CMF C7 H16 O4



L8 ANSWER 3 OF 3 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2004:427714 HCAPLUS

DN 141:9606

TI Boron-containing compound, ion-conductive polymer and polyelectrolyte for

electrochemical devices
 IN Okumura, Takefumi; Nishimura, Shin; Iwayasu,
 Norio; Yokoyama, Shoichi; Yabe, Takeshi
 PA Hitachi, Ltd., Japan; NOF Corporation
 SO Eur. Pat. Appl., 25 pp.
 CODEN: EPXXDW

DT Patent
 LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP 1422781	A1	20040526	EP 2003-26140	20031113
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	CN 1502644	A	20040609	CN 2003-10118012	20031120
	US 2004147697	A1	20040729	US 2003-717645	20031121
PRAI	JP 2002-337789	A	20021121		
	EP 2003-13841	A	20030618		

OS MARPAT 141:9606

AB An object of the present invention is to provide a boron-containing compound capable of forming an ion-conductive polyelectrolyte having high ion-conductive properties, and a polymer of the compound. According to the present invention, there are provided a polymerizable boron-containing compound of formula Z1(AO)pOB(O(AO)mZ2)O(AO)nZ3 [where B is boron atom; Z1, Z2, and Z3 are organic groups having an acryloyl or methacryloyl group; AOs are independently an oxyalkylene group of C1-6 and are of one or more kinds; and m, n and p are independently an average number of moles of the oxyalkylene group(s) added of <4 and >0, provided that m+n+p ≥1] a polymer thereof, a polymer of a compound of formula Z4(AO)p1OB(O(AO)m1Z5)O(AO)n1Z6 and a compound of formula R1(AO)p2OB(O(AO)m2R2)O(AO)n2R3 [where Z4, Z5, and Z6 is an organic group having an acryloyl or methacryloyl group; R1, R2 and R3 are independently a hydrocarbon group of C1-10; AOs are independently an oxyalkylene group of C1-6 and are of one or more kinds; and m1, n1, p1, m2, n2, and p2 are independently an average no. of moles of the oxyalkylene group(s) added of <4 and >0, provided that each of the sum of m1+n1+p1 and the sum of m2+n2+p3 ≥1] and a polyelectrolyte for electrochem. device comprising either of these polymers and at least one electrolyte salt:.

IT 693782-27-3P 693782-28-4P 693782-29-5P
 693782-30-8P 693782-31-9P 693782-32-0P

RL: DEV (Device component use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)

(boron-containing compound, ion-conductive polymer and polyelectrolyte for electrochem. devices)

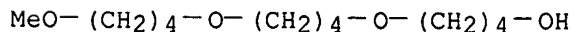
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CN Boric acid (H3BO3), 4-[4-(4-methoxybutoxy)butoxy]butyl
 4-[4-[(2-methyl-1-oxo-2-propenyl)oxy]butoxy]butyl ester (9CI) (CA INDEX NAME)

CM 1

CRN 693782-26-2

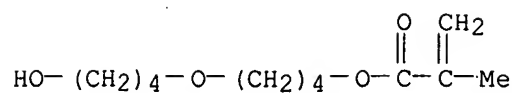
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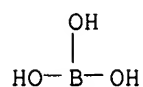
CMF C12 H22 O4



CM 3

CRN 10043-35-3

CMF B H3 O3



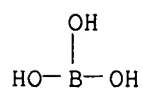
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CN Boric acid (H3BO3), 2-[2-(2-methoxyethoxy)ethoxy]ethyl
 2-[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]ethyl ester (9CI) (CA INDEX
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CM 1

CRN 10043-35-3

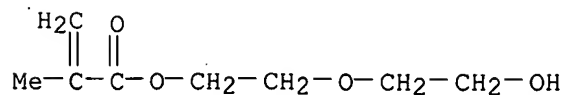
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CM 2

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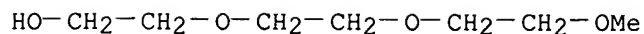
CMF C8 H14 O4



CM 3

CRN 112-35-6

CMF C7 H16 O4



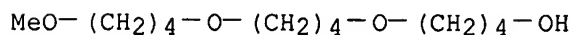
RN 693782-29-5 HCAPLUS
 CN 2-Propenoic acid, 2-methyl-, 4-(4-hydroxybutoxy)butyl ester, ester with boric acid (H3BO3) 4-[4-(4-methoxybutoxy)butoxy]butyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 693782-27-3
 CMF C13 H28 O4 . x C12 H22 O4 . x B H3 O3

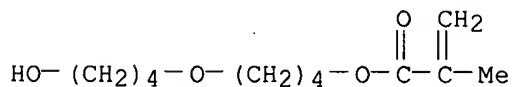
CM 2

CRN 693782-26-2
 CMF C13 H28 O4



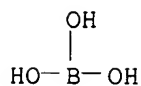
CM 3

CRN 78972-17-5
 CMF C12 H22 O4



CM 4

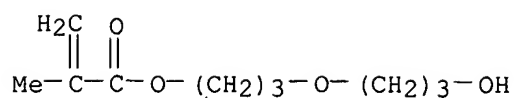
CRN 10043-35-3
 CMF B H3 O3



RN 693782-30-8 HCAPLUS
 CN Boric acid (H3BO3), 3-[3-(3-methoxypropoxy)propoxy]propyl 3-[3-[(2-methyl-1-oxo-2-propenyl)oxy]propoxy]propyl ester (9CI) (CA INDEX NAME)

CM 1

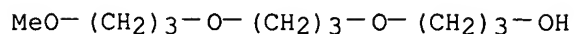
CRN 78972-16-4
 CMF C10 H18 O4



CM 2

CRN 13133-29-4

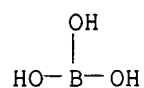
CMF C10 H22 O4



CM 3

CRN 10043-35-3

CMF B H3 O3



RN 693782-31-9 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 3-(3-hydroxypropoxy)propyl ester, ester with boric acid (H3BO3) 3-[3-(3-methoxypropoxy)propoxy]propyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

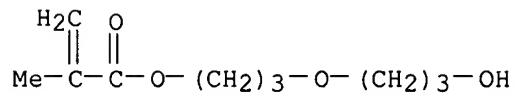
CRN 693782-30-8

CMF C10 H22 O4 . x C10 H18 O4 . x B H3 O3

CM 2

CRN 78972-16-4

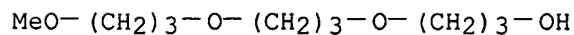
CMF C10 H18 O4



CM 3

CRN 13133-29-4

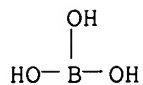
CMF C10 H22 O4



CM 4

CRN 10043-35-3

CMF B H3 O3



RN 693782-32-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(2-hydroxyethoxy)ethyl ester, ester with boric acid (H3BO3) 2-[2-(2-methoxyethoxy)ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

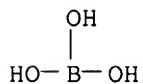
CRN 693782-28-4

CMF C8 H14 O4 . x C7 H16 O4 . x B H3 O3

CM 2

CRN 10043-35-3

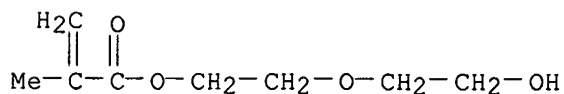
CMF B H3 O3



CM 3

CRN 2351-43-1

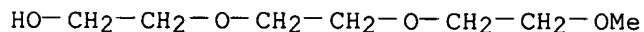
CMF C8 H14 O4



CM 4

CRN 112-35-6

CMF C7 H16 O4



=> fil reg

FILE 'REGISTRY' ENTERED AT 09:16:48 ON 30 OCT 2006

USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.

PLEASE SEE "HELP USAGETERMS" FOR DETAILS.

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Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 29 OCT 2006 HIGHEST RN 911424-89-0

DICTIONARY FILE UPDATES: 29 OCT 2006 HIGHEST RN 911424-89-0

New CAS Information Use Policies, enter HELP USAGETERMS for details.

TSCA INFORMATION NOW CURRENT THROUGH June 30, 2006

Please note that search-term pricing does apply when conducting SmartSELECT searches.

REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

<http://www.cas.org/ONLINE/UG/regprops.html>

=> d ide can tot 14

L4 ANSWER 1 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN

RN 693782-32-0 REGISTRY

ED Entered STN: 16 Jun 2004

CN 2-Propenoic acid, 2-methyl-, 2-(2-hydroxyethoxy)ethyl ester, ester with boric acid (H3BO3) 2-[2-(2-methoxyethoxy)ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

MF (C8 H14 O4 . x C7 H16 O4 . x B H3 O3)x

CI PMS

PCT Polyacrylic, Polyether

SR CA

LC STN Files: CA, CAPLUS, USPATFULL

CM 1

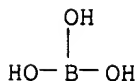
CRN 693782-28-4

CMF C8 H14 O4 . x C7 H16 O4 . x B H3 O3

CM 2

CRN 10043-35-3

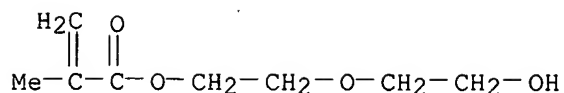
CMF B H3 O3



CM 3

CRN 2351-43-1

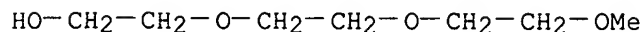
CMF C8 H14 O4



CM 4

CRN 112-35-6

CMF C7 H16 O4



3 REFERENCES IN FILE CA (1907 TO DATE)

3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 141:9609

REFERENCE 2: 141:9608

REFERENCE 3: 141:9606

L4 ANSWER 2 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN

RN 693782-31-9 REGISTRY

ED Entered STN: 16 Jun 2004

CN 2-Propenoic acid, 2-methyl-, 3-(3-hydroxypropoxy)propyl ester, ester with
boric acid (H3BO3) 3-[3-(3-methoxypropoxy)propoxy]propyl ester,
homopolymer (9CI) (CA INDEX NAME)

MF (C10 H22 O4 . x C10 H18 O4 . x B H3 O3)x

CI PMS

PCT Polyacrylic, Polyether

SR CA

LC STN Files: CA, CAPLUS, USPATFULL

CM 1

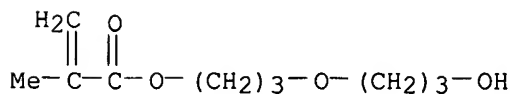
CRN 693782-30-8

CMF C10 H22 O4 . x C10 H18 O4 . x B H3 O3

CM 2

CRN 78972-16-4

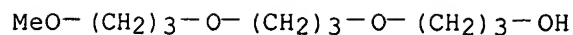
CMF C10 H18 O4



CM 3

CRN 13133-29-4

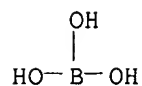
CMF C10 H22 O4



CM 4

CRN 10043-35-3

CMF B H3 O3



3 REFERENCES IN FILE CA (1907 TO DATE)
3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 141:9609

REFERENCE 2: 141:9608

REFERENCE 3: 141:9606

L4 ANSWER 3 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN

RN 693782-30-8 REGISTRY

ED Entered STN: 16 Jun 2004

CN Boric acid (H3BO3), 3-[3-(3-methoxypropoxy)propoxy]propyl
3-[3-[(2-methyl-1-oxo-2-propenyl)oxy]propoxy]propyl ester (9CI) (CA INDEX
NAME)

MF C10 H22 O4 . x C10 H18 O4 . x B H3 O3

CI COM

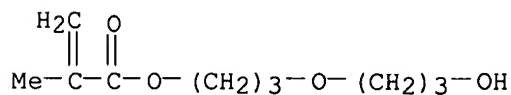
SR CA

LC STN Files: CA, CAPLUS, USPATFULL .

CM 1

CRN 78972-16-4

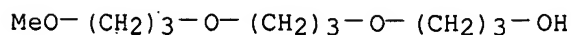
CMF C10 H18 O4 .



CM 2

CRN 13133-29-4

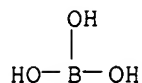
CMF C10 H22 O4 .



CM 3

CRN 10043-35-3

CMF B H3 O3



3 REFERENCES IN FILE CA (1907 TO DATE)
3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 141:9609

REFERENCE 2: 141:9608

REFERENCE 3: 141:9606

L4 ANSWER 4 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN
RN 693782-29-5 REGISTRY
ED Entered STN: 16 Jun 2004
CN 2-Propenoic acid, 2-methyl-, 4-(4-hydroxybutoxy)butyl ester, ester with
boric acid (H3BO3) 4-[4-(4-methoxybutoxy)butoxy]butyl ester, homopolymer
(9CI) (CA INDEX NAME)
MF (C13 H28 O4 . x C12 H22 O4 . x B H3 O3)x
CI PMS
PCT Polyacrylic, Polyether
SR CA
LC STN Files: CA, CAPLUS, USPATFULL

CM 1

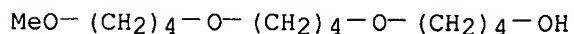
CRN 693782-27-3

CMF C13 H28 O4 . x C12 H22 O4 . x B H3 O3

CM 2

CRN 693782-26-2

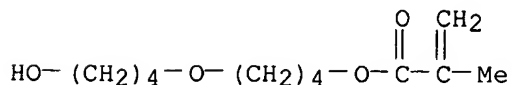
CMF C13 H28 O4



CM 3

CRN 78972-17-5

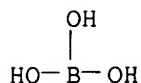
CMF C12 H22 O4



CM 4

CRN 10043-35-3

CMF B H3 O3



3 REFERENCES IN FILE CA (1907 TO DATE)
3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 141:9609

REFERENCE 2: 141:9608

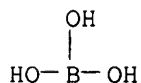
REFERENCE 3: 141:9606

L4 ANSWER 5 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN
RN 693782-28-4 REGISTRY
ED Entered STN: 16 Jun 2004
CN Boric acid (H3BO3), 2-[2-(2-methoxyethoxy)ethoxy]ethyl
2-[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]ethyl ester (9CI) (CA INDEX
NAME)
MF C8 H14 O4 . x C7 H16 O4 . x B H3 O3
CI COM
SR CA
LC STN Files: CA, CAPLUS, USPATFULL

CM 1

CRN 10043-35-3

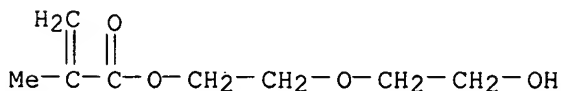
CMF B H3 O3



CM 2

CRN 2351-43-1

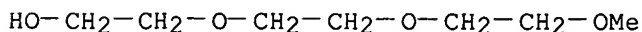
CMF C8 H14 O4



CM 3

CRN 112-35-6

CMF C7 H16 O4



3 REFERENCES IN FILE CA (1907 TO DATE)
3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 141:9609

REFERENCE 2: 141:9608

REFERENCE 3: 141:9606

L4 ANSWER 6 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN
RN 693782-27-3 REGISTRY
ED Entered STN: 16 Jun 2004
CN Boric acid (H3BO3), 4-[4-(4-methoxybutoxy)butoxy]butyl
4-[4-[(2-methyl-1-oxo-2-propenyl)oxy]butoxy]butyl ester (9CI) (CA INDEX
NAME)
MF C13 H28 O4 . x C12 H22 O4 . x B H3 O3
CI COM
SR CA
LC STN Files: CA, CAPLUS, USPATFULL

CM 1

CRN 693782-26-2
CMF C13 H28 O4

MeO- (CH2)4-O- (CH2)4-O- (CH2)4-OH

CM 2

CRN 78972-17-5
CMF C12 H22 O4

HO- (CH2)4-O- (CH2)4-O- $\begin{array}{c} \text{O} \quad \text{CH}_2 \\ \parallel \quad \parallel \\ \text{C} - \text{C} - \text{Me} \end{array}$

CM 3

CRN 10043-35-3
CMF B H3 O3

$\begin{array}{c} \text{OH} \\ | \\ \text{HO} - \text{B} - \text{OH} \end{array}$

3 REFERENCES IN FILE CA (1907 TO DATE)
3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 141:9609

REFERENCE 2: 141:9608

REFERENCE 3: 141:9606

=> d his

(FILE 'HOME' ENTERED AT 09:13:05 ON 30 OCT 2006)
SET COST OFF

FILE 'HCAPLUS' ENTERED AT 09:13:16 ON 30 OCT 2006
L1 2 S (US20040101758 OR US20020101759)/PN OR (US2003-717646# OR US2
SEL RN

FILE 'REGISTRY' ENTERED AT 09:14:00 ON 30 OCT 2006
L2 20 S E1-E20
L3 7 S L2 AND B/ELS NOT RIS/CI
L4 6 S L3 NOT LI/ELS

FILE 'HCAPLUS' ENTERED AT 09:14:49 ON 30 OCT 2006
L5 3 S L4
L6 3 S L5 AND (OKUMURA? OR TAKEFUMI? OR NISHIMURA? OR SHIN? OR IWAYA
L7 1 S L5 AND HITACHI?/PA,CS
L8 3 S L5-L7
L9 1 S L8 NOT L1
SEL RN

FILE 'REGISTRY' ENTERED AT 09:15:54 ON 30 OCT 2006
L10 16 S E21-E36
L11 0 S L10 AND B/ELS NOT L2

FILE 'HCAPLUS' ENTERED AT 09:16:33 ON 30 OCT 2006

FILE 'REGISTRY' ENTERED AT 09:16:48 ON 30 OCT 2006

=> => fil reg

FILE 'REGISTRY' ENTERED AT 11:27:31 ON 30 OCT 2006
USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.
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STRUCTURE FILE UPDATES: 29 OCT 2006 HIGHEST RN 911424-89-0
DICTIONARY FILE UPDATES: 29 OCT 2006 HIGHEST RN 911424-89-0

New CAS Information Use Policies, enter HELP USAGETERMS for details.

TSCA INFORMATION NOW CURRENT THROUGH June 30, 2006

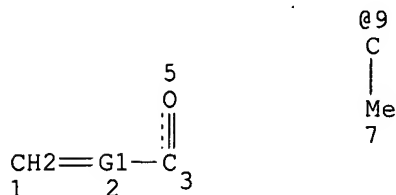
Please note that search-term pricing does apply when
conducting SmartSELECT searches.

REGISTRY includes numerically searchable data for experimental and
predicted properties as well as tags indicating availability of
experimental property data in the original document. For information
on property searching in REGISTRY, refer to:

<http://www.cas.org/ONLINE/UG/regprops.html>

=> d sta que 18

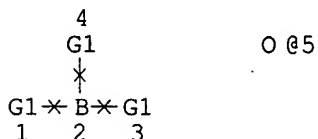
L1 387535 SEA FILE=REGISTRY ABB=ON PLU=ON B/ELS OR (?BORON? OR ?BORIC?
OR ?BORAT?)/CNS
L2 STR



VAR G1=CH/9
NODE ATTRIBUTES:
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 6

STEREO ATTRIBUTES: NONE
L4 1182 SEA FILE=REGISTRY SUB=L1 SSS FUL L2
L6 STR



VAR G1=5/X
NODE ATTRIBUTES:
NSPEC IS RC AT 5
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 5

STEREO ATTRIBUTES: NONE
L8 340 SEA FILE=REGISTRY SUB=L4 SSS FUL L6

100.0% PROCESSED 1180 ITERATIONS 340 ANSWERS
SEARCH TIME: 00.00.01

=> d his

(FILE 'HOME' ENTERED AT 10:22:35 ON 30 OCT 2006)
SET COST OFF

FILE 'REGISTRY' ENTERED AT 10:22:50 ON 30 OCT 2006
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L2 STR
L3 50 S L2 SAM SUB=L1
L4 1182 S L2 FUL SUB=L1
SAV TEMP L4 LAURA717/A

L5 STR
L6 STR L5
L7 18 S L6 SAM SUB=L4
L8 340 S L6 FUL SUB=L4
SAV TEMP L8 LAURA717A/A
L9 63 S BH3O3 AND L8
L10 25 S L9 AND 4/ELC.SUB
L11 23 S L10 NOT C6/ES
L12 19 S L11 NOT (C6H10O4 OR C10H18O4)
L13 4 S L11 NOT L12
L14 2 S L13 AND 2/NC
L15 2 S L13 NOT L14
L16 19 S L11 NOT L13
L17 17 S L16 NOT (57-55-6 OR 5919-74-4)/CRN
L18 4 S L10 NOT L16,L15
L19 19 S L15,L17
L20 38 S L9 NOT L10
L21 277 S L8 NOT L9
L22 101 S L21 AND 1/NC
L23 6 S L22 AND (C15H21BO9 OR C18H27BO9 OR C24H39BO12)
L24 9 S L22 AND C2H4O
SEL RN 1 8 9
L25 6 S L24 NOT E16-E18
L26 18 S L19 NOT C14H24O4
L27 176 S L21 NOT L22
L28 56 S L27 NOT BF4
L29 30 S L23,L25,L26
SAV TEMP L29 LAURA717B/A
L30 6 S 693782-27-3 OR 693782-28-4 OR 693782-29-5 OR 693782-30-8 OR 6
L31 24 S L29 NOT L30

FILE 'HCAOLD' ENTERED AT 11:24:32 ON 30 OCT 2006

L32 0 S L31

FILE 'HCAPLUS' ENTERED AT 11:24:34 ON 30 OCT 2006

L33 13 S L31
L34 10 S L33 AND (PY<=2003 OR PRY<=2003 OR AY<=2003)
L35 7 S L33 AND (OKUMURA? OR TAKEFUMI? OR NISHIMURA? OR SHIN? OR IWAY
L36 3 S L33 AND HITACHI?/PA,CS
L37 4 S L34 AND L35,L36
L38 10 S L34,L37
L39 3 S L33-L37 NOT L38

FILE 'USPATFULL' ENTERED AT 11:27:16 ON 30 OCT 2006

L40 2 S L31

FILE 'REGISTRY' ENTERED AT 11:27:31 ON 30 OCT 2006

=> fil uspatful

FILE 'USPATFULL' ENTERED AT 11:27:48 ON 30 OCT 2006

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FILE COVERS 1971 TO PATENT PUBLICATION DATE: 26 Oct 2006 (20061026/PD)

FILE LAST UPDATED: 26 Oct 2006 (20061026/ED)

HIGHEST GRANTED PATENT NUMBER: US7127745

HIGHEST APPLICATION PUBLICATION NUMBER: US2006242744

CA INDEXING IS CURRENT THROUGH 24 Oct 2006 (20061024/UPCA)

ISSUE CLASS FIELDS (/INCL) CURRENT THROUGH: 26 Oct 2006 (20061026/PD)

REVISED CLASS FIELDS (/NCL) LAST RELOADED: Jun 2006

USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Jun 2006

=> d 140 bib abs hitstr tot

L40 ANSWER 1 OF 2 USPATFULL on STN
AN 2004:335886 USPATFULL
TI Process for producing boric ester compound, electrolyte for
electrochemical device, and secondary battery
IN Yokoyama, Shoichi, Kanagawa, JAPAN
Yabe, Takeshi, Kanagawa, JAPAN
PI US 2004266981 A1 20041230
US 6998465 B2 20060214
AI US 2004-489418 A1 20040312 (10)
WO 2002-JP10049 20020927
PRAI JP 2001-301122 20010928
JP 2002-98060 20020329
DT Utility
FS APPLICATION
LREP ANTONELLI, TERRY, STOUT & KRAUS, LLP, 1300 NORTH SEVENTEENTH STREET,
SUITE 1800, ARLINGTON, VA, 22209-9889
CLMN Number of Claims: 16
ECL Exemplary Claim: 1
DRWN No Drawings
LN.CNT 1634
CAS INDEXING IS AVAILABLE FOR THIS PATENT.
AB A process for producing a boric acid ester compound which comprises
esterifying a compound represented by the formula (1):

X--[O(AO).sub.n--H].sub.a (1)

wherein X represents a group independently selected from a residue of a
compound having 1 to 6 hydroxyl groups, an acryloyl group and a
methacryloyl group with a boron-containing compound represented by the
formula (2):

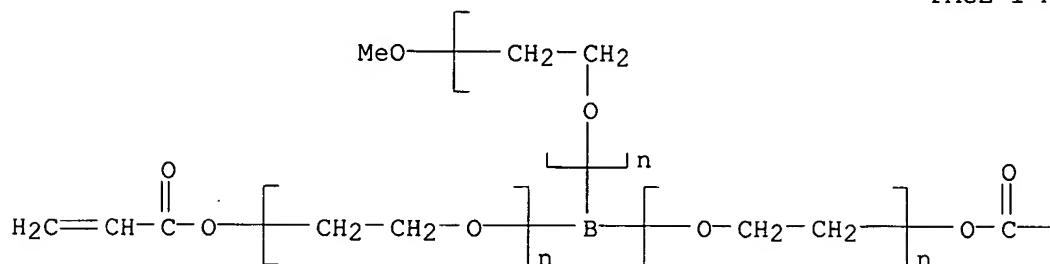
(RO).sub.3--B (2)

wherein R represents an alkyl group having 1 to 4 carbon atoms. The
present invention can provide a boric acid ester compound which has a
high ion conductivity, which is useful as a material for an
electrochemical device, such as a secondary battery or a capacitor,
having excellent safety and which is low in water and impurity contents,
a polymer electrolyte containing the boric acid ester compound, and a
secondary battery using the polymer electrolyte.

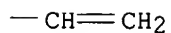
CAS INDEXING IS AVAILABLE FOR THIS PATENT.
IT 512206-26-7P 512206-27-8P 512206-28-9P
512206-30-3P 512206-31-4P 512206-32-5P
512776-99-7P 512777-00-3P
(manufacture of borate ester compds. for secondary battery electrolytes)
RN 512206-26-7 USPATFULL
CN Poly(oxy-1,2-ethanediyl), α,α',α'' -
borylidynetris[ω -(2-methyl-1-oxo-2-propenyl)oxy]- (9CI) (CA
INDEX NAME)

RN 512206-28-9 USPATFULL
 CN Poly(oxy-1,2-ethanediyl), ω -methoxy- ω' , ω'' -bis[(1-oxo-2-propenyl)oxy]- α , α' , α'' -borylidynetris- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 1-B



RN 512206-30-3 USPATFULL
 CN Poly(oxy-1,2-ethanediyl), α , α' , α'' -borylidynetris[ω -[(1-oxo-2-propenyl)oxy]-, homopolymer (9CI) (CA INDEX NAME)

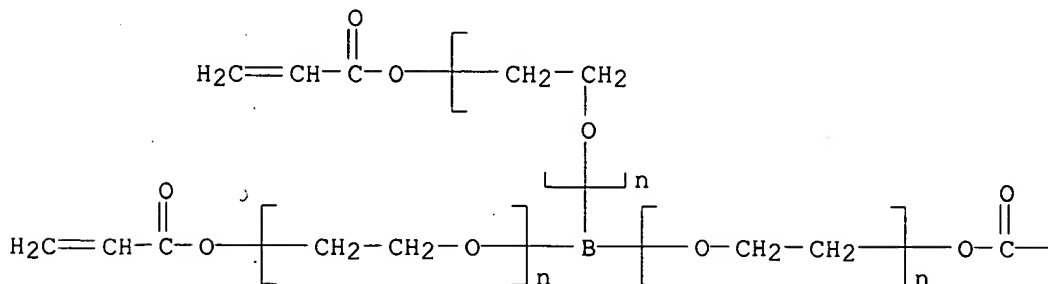
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CRN 512206-27-8

CMF (C2 H4 O)_n (C2 H4 O)_n (C2 H4 O)_n C9 H9 B O6

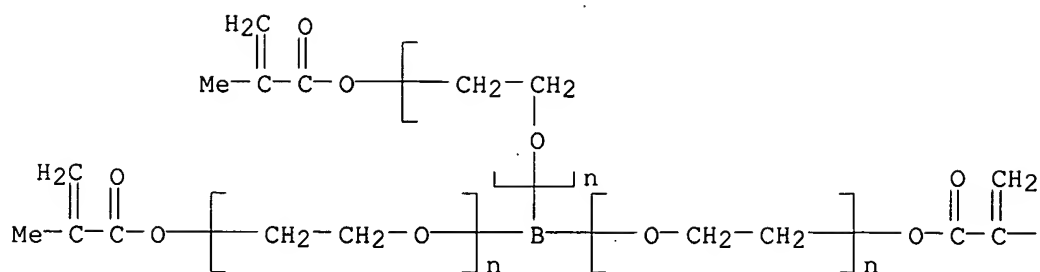
CCI PMS

PAGE 1-A



CCI PMS

PAGE 1-A



PAGE 1-B

— Me

RN 512776-99-7 USPATFULL

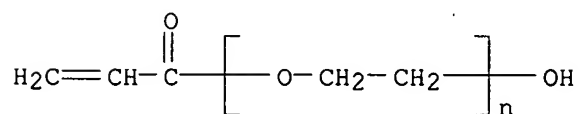
CN Oxirane, methyl-, polymer with oxirane, monomethyl ether, ester with boric acid (H3BO3) ester with α -(1-oxo-2-propenyl)- ω -hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 26403-58-7

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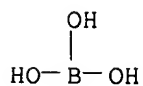
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98	98
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100	100



CM 2

CRN 10043-35-3

CMF B H3 O3



CM 3

CRN 9063-06-3

CMF (C3 H6 O . C2 H4 O)x . C H4 O

CDES 8:GD,ETHER

CM 4

CRN 67-56-1

CMF C H4 O

H₃C-OH

CM 5

CRN 9003-11-6

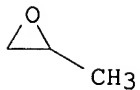
CMF (C3 H6 O . C2 H4 O)x

CCI PMS

CM 6

CRN 75-56-9

CMF C3 H6 O



CM 7

CRN 75-21-8

CMF C2 H4 O



RN 512777-00-3 USPATFULL

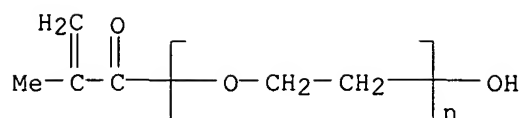
CN Poly(oxy-1,2-ethanediyl), α -(2-methyl-1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H₃BO₃) ester with α -hydro- ω -hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 25736-86-1

CMF (C2 H4 O)_n C4 H6 O₂

CCI PMS

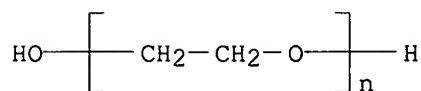


CM 2

CRN 25322-68-3

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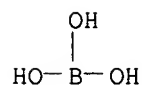
CCI PMS



CM 3

CRN 10043-35-3

CMF B H3 O3



L40 ANSWER 2 OF 2 USPATFULL on STN

AN 2004:323217 USPATFULL

TI Electrolyte for secondary battery and secondary battery

IN Yokoyama, Shoichi, Yokohama, JAPAN

Wakihara, Masataka, Yokohama, JAPAN

Kobayashi, Takao, Mihara-gun, JAPAN

Suwa, Kentaro, Oota-ku, JAPAN

PA NOF Corporation, Tokyo, JAPAN (non-U.S. corporation)

PI US 6833220 B1 20041221

WO 2001039316 20010531

AI US 2002-130952 20020524 (10)

WO 2000-JP8254 20001122

20020524 PCT 371 date

PRAI JP 1999-332586 19991124

JP 2000-87754 20000328

DT Utility

FS GRANTED

EXNAM Primary Examiner: Ryan, Patrick; Assistant Examiner: Dove, Tracy

LREP Sughrue Mion, PLLC

CLMN Number of Claims: 32

ECL Exemplary Claim: 1

DRWN 0 Drawing Figure(s); 0 Drawing Page(s)

LN.CNT 1823

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB The invention provides an electrolyte for secondary battery having a high ionic conductivity and an excellent safety and a secondary battery having an excellent cycle life performance comprising such an

electrolyte.

In other words, the invention lies in an electrolyte for secondary battery comprising an ionic compound and an organic polymer compound, wherein the organic polymer compound comprises a compound represented by the general formula (1) or a boric acid ester compound obtained by the esterification of the compound represented by the general formula (1) with boric acid or boric anhydride:

Z.sup.1--[(A.sup.10).sub.1--R.sup.1].sub.a (1)

wherein Z.sup.1 represents a residue of compound having from 1 to 6 hydroxyl groups; A.sup.10 represents one or a mixture of two or more of C.sub.2-C.sub.4 oxyalkylene groups; R.sup.1 represents a group selected from the group consisting of cyanoethyl group, C.sub.1-C.sub.12 hydrocarbon group and hydrogen atom; 1 represents an integer of from 0 to 600; and the suffix a represents an integer of from 1 to 6, with the proviso that 1a ranges from 0 to 600, and a secondary battery comprising the electrolyte for secondary battery,

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

IT 340814-65-5 340814-66-6 340814-67-7

(compns. of oxyalkylene polymer electrolytes for secondary lithium batteries)

RN 340814-65-5 USPATFULL

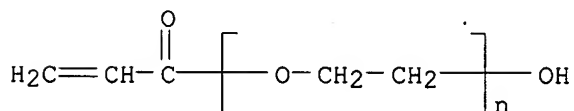
CN Poly(oxy-1,2-ethanediyl), α -(1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H₃BO₃) ester with α -methyl- ω -hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 26403-58-7

CMF (C2 H4 O)_n C3 H4 O2

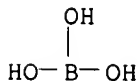
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CM 2

CRN 10043-35-3

CMF B H3 O3

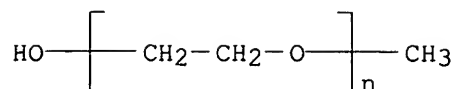


CM 3

CRN 9004-74-4

CMF (C2 H4 O)_n C H4 O

CCI PMS



RN 340814-66-6 USPATFULL

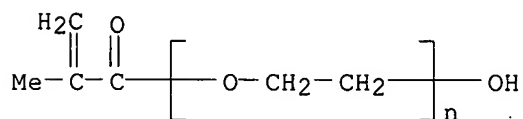
CN Poly(oxy-1,2-ethanediyl), α -(2-methyl-1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H₃BO₃) (9CI) (CA INDEX NAME)

CM 1

CRN 25736-86-1

CMF (C₂ H₄ O)_n C₄ H₆ O₂

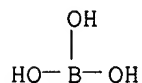
CCI PMS



CM 2

CRN 10043-35-3

CMF B H₃ O₃



RN 340814-67-7 USPATFULL

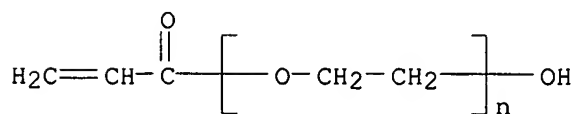
CN Poly(oxy-1,2-ethanediyl), α -(1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H₃BO₃) (9CI) (CA INDEX NAME)

CM 1

CRN 26403-58-7

CMF (C₂ H₄ O)_n C₃ H₄ O₂

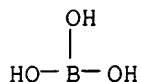
CCI PMS



CM 2

CRN 10043-35-3

CMF B H₃ O₃



=> fil hcaplus

FILE 'HCAPLUS' ENTERED AT 11:28:00 ON 30 OCT 2006

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FILE COVERS 1907 - 30 Oct 2006 VOL 145 ISS 19

FILE LAST UPDATED: 29 Oct 2006 (20061029/ED)

New CAS Information Use Policies, enter HELP USAGETERMS for details.

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d 138 bib abs hitstr retable tot

L38 ANSWER 1 OF 10 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2003:694134 HCAPLUS

DN 139:232985

TI Polymer solid electrolyte and polymer solid electrolyte battery

IN Bando, Toshinori; Kuratomi, Junichi; Ono, Tetsuo

PA Yuasa Corporation, Japan

SO Jpn. Kokai Tokkyo Koho, 11 pp.

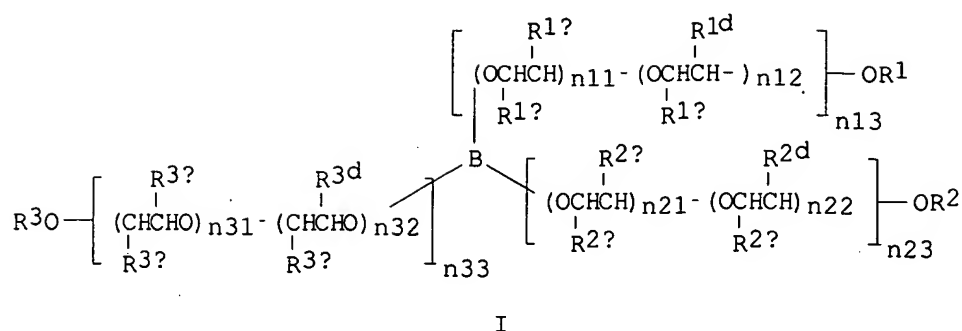
CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	-----	----	-----	-----	-----
PI	JP 2003249266	A2	<u>20030905</u>	JP 2002-48481	20020225 <--
PRAI	JP 2002-48481		20020225	<--	
OS	MARPAT 139:232985				
GI					



AB The electrolyte contains an electrolyte salt and a polymer; where the polymer has repeating structure units derived from a compound I [$R_1 = C>1$ nonpolymerizable functional group; $R_2, R_3 =$ polymerizable functional group; $R_{1a}, R_{1b}, R_{1c}, R_{1d}, R_{2a}, R_{2b}, R_{2c}, R_{2d}, R_{3a}, R_{3b}, R_{3c}, R_{3d} = H$ or $C1-3$ alkyl group; $n_{11}, n_{12}, n_{13}, n_{21}, n_{22}, n_{23}, n_{31}, n_{32}, n_{33} =$ integer $0-100$; $(n_{21} + n_{22} + n_{23}) \leq 0$; $(n_{31} + n_{32} + n_{33}) \leq 0$; $n_{13}(n_{11}+n_{12}) > n_{23}(n_{21}+n_{22}) > n_{33}(n_{31}+n_{32})$]. The battery has the above electrolyte, a cathode containing a transition metal oxide based active mass and an anode containing a Li alloy, Li, or Li-intercalating substance based anode material.

IT 512206-28-9

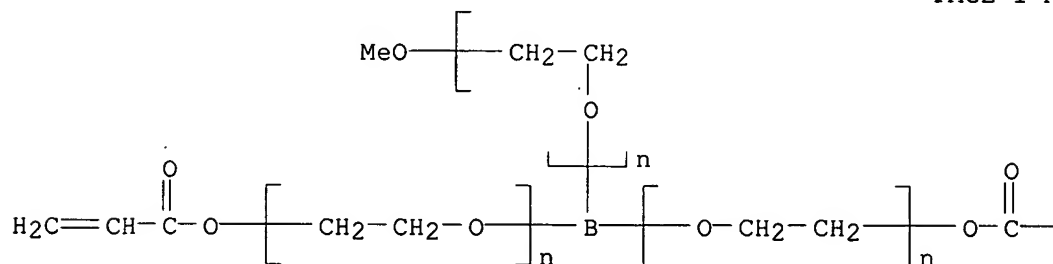
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RL: DEV (Device component use); USES (Uses)
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(solid electrolytes containing electrolyte salts and polymers for secondary lithium batteries)

RN 512206-28-9 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), ω -methoxy- ω' , ω'' -bis[(1-oxo-2-propenyl)oxy]- α , α' , α'' -borylidynetris- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 1-B

$$-\text{CH}=\text{CH}_2$$

L38 ANSWER 2 OF 10 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2003:301090 HCAPLUS

DN 138:324031

TI Manufacture of borate ester compound, electrolyte for electrochemical device, and secondary battery

IN Yokoyama, Shoichi; Yabe, Takeshi

PA NOF Corporation, Japan

SO PCT Int. Appl., 55 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE	
PI	WO 2003031453	A1	20030417	WO 2002-JP10049	20020927 <--	
	W:			AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW		
	RW:			GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG		
	JP 2003201344	A2	20030718	JP 2002-282068	20020927 <--	
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	CN 1596260	A	20050316	CN 2002-823808	20020927 <--	
	JP 2004002342	A2	20040108	JP 2003-82497	20030325 <--	
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PRAI	JP 2001-301122	A	20010928	<--		
	JP 2002-98060	A	20020329	<--		
	WO 2002-JP10049	W	20020927	<--		

AB The ester compound is prepared by esterification of a compound I X[O(AO)nH]a (X = residue of a compound having 1-6 OH groups; AO = C2-4 oxyalkylene group; n = 0-600; a = 1-6) with a B containing compound II (RO)3B (R = C1-4 alkyl group).

The battery uses an electrolyte containing the borate ester compound or its copolymer.

IT 512206-26-7P 512206-27-8P 512206-28-9P

512206-30-3P 512206-31-4P 512206-32-5P

512776-99-7P 512777-00-3P

RL: DEV (Device component use); IMF (Industrial manufacture); PREP

(Preparation); USES (Uses)

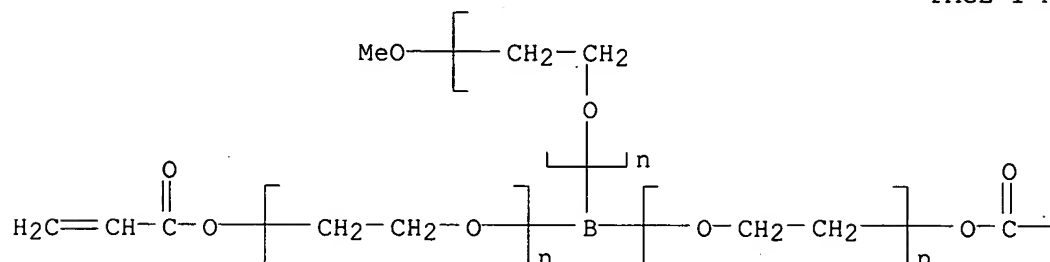
(manufacture of borate ester compds. for secondary battery electrolytes)

RN 512206-26-7 HCAPLUS

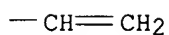
CN Poly(oxy-1,2-ethanediyl), $\alpha, \alpha', \alpha''$ -borylidynetris[ω -[(2-methyl-1-oxo-2-propenyl)oxy]- (9CI) (CA INDEX NAME)

RN 512206-28-9 HCAPLUS
 CN Poly(oxy-1,2-ethanediyl), ω -methoxy- ω' , ω'' -bis[(1-oxo-2-propenyl)oxy]- α , α' , α'' -borylidynetris- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 1-B



RN 512206-30-3 HCAPLUS
 CN Poly(oxy-1,2-ethanediyl), α , α' , α'' -borylidynetris[ω -[(1-oxo-2-propenyl)oxy]-, homopolymer (9CI) (CA INDEX NAME)

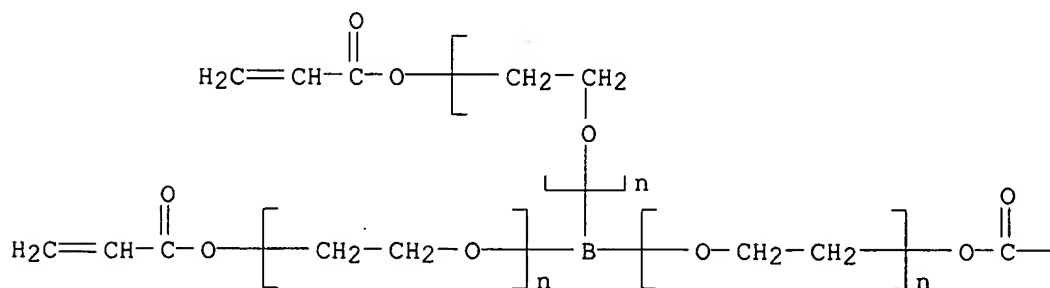
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CRN 512206-27-8

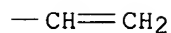
CMF (C2 H4 O) $_n$ (C2 H4 O) $_n$ (C2 H4 O) $_n$ C9 H9 B O6

CCI PMS

PAGE 1-A



PAGE 1-B



RN 512206-31-4 HCAPLUS
 CN Poly(oxy-1,2-ethanediyl), ω -methoxy- ω' , ω'' -bis[(1-oxo-2-propenyl)oxy]- α,α',α'' -borylidynetris-, homopolymer
 (9CI) (CA INDEX NAME)

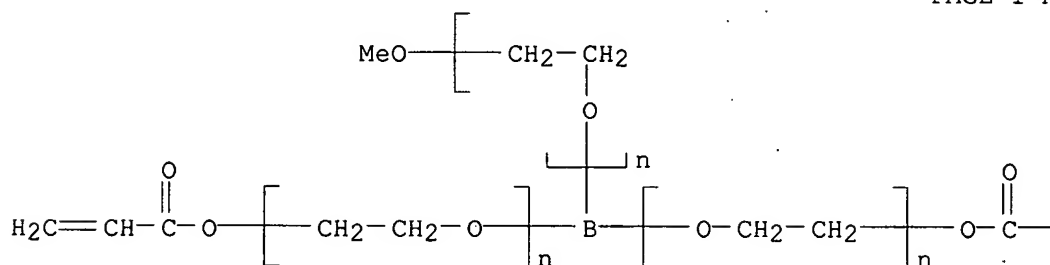
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CRN 512206-28-9

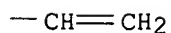
CMF (C2 H4 O)_n (C2 H4 O)_n (C2 H4 O)_n C7 H9 B O5

CCI PMS

PAGE 1-A



PAGE 1-B



RN 512206-32-5 HCAPLUS
 CN Poly(oxy-1,2-ethanediyl), α,α',α'' -borylidynetris[ω -[(2-methyl-1-oxo-2-propenyl)oxy]-, homopolymer
 (9CI) (CA INDEX NAME)

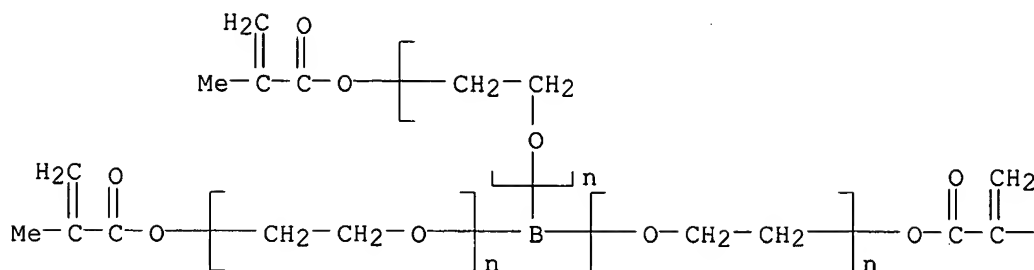
CM 1

CRN 512206-26-7

CMF (C2 H4 O)_n (C2 H4 O)_n (C2 H4 O)_n C12 H15 B O6

CCI PMS .

PAGE 1-A



PAGE 1-B

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RN 512776-99-7 HCAPLUS

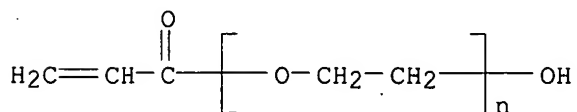
CN Oxirane, methyl-, polymer with oxirane, monomethyl ether, ester with boric acid (H3BO3) ester with α -(1-oxo-2-propenyl)- ω -hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 26403-58-7

$$\text{CMF} \quad (\text{C}_2 \text{ H}_4 \text{ O})_n \text{ C}_3 \text{ H}_4 \text{ O}_2$$

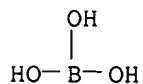
CCI PMS



CM 2

CRN 10043-35-3

CMF B H3 O3



CM 3

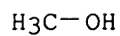
CRN 9063-06-3

CMF (C3 H6 O . C2 H4 O)x . C H4 O

CM 4

CRN 67-56-1

CMF C H4 O



CM 5

CRN 9003-11-6

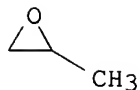
CMF (C3 H6 O . C2 H4 O)x

CCI PMS

CM 6

CRN 75-56-9

CMF C3 H6 O



CM 7

CRN 75-21-8

CMF C2 H4 O



RN 512777-00-3 HCAPLUS

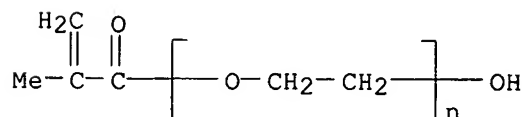
CN Poly(oxy-1,2-ethanediyl), α -(2-methyl-1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H3BO3) ester with α -hydro- ω -hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 25736-86-1

CMF (C2 H4 O)n C4 H6 O2

CCI PMS

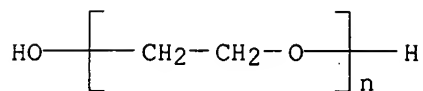


CM 2

CRN 25322-68-3

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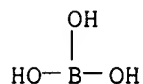
CCI PMS



CM 3

CRN 10043-35-3

CMF B H3 O3



RETABLE

Referenced Author (RAU)	Year (RPY)	VOL (RVL)	PG (RPG)	Referenced Work (RWK)	Referenced File
Dai - Ichi Kogyo Seiyak	2001			WO 0118094 A1	HCAPLUS
Dai - Ichi Kogyo Seiyak	2001			EP 1160268 A1	HCAPLUS
Dai - Ichi Kogyo Seiyak	2001			JP 200172875 A	
Dai - Ichi Kogyo Seiyak	2001			JP 200172876 A	
Dai - Ichi Kogyo Seiyak	2001			JP 200172877 A	
Mine Safety Appliances	1972			JP 47-29323 A	
Nof Corp	2001			WO 0139316 A1	HCAPLUS
Nof Corp	2001			EP 1258938 A1	HCAPLUS
Nof Corp	2001			JP 2001155771 A	HCAPLUS
Nof Corp	2001			JP 2001273925 A	HCAPLUS
Nof Corp	2002			JP 2002158039 A	HCAPLUS
Nof Corp	2002			JP 2002348323 A	HCAPLUS

L38 ANSWER 3 OF 10 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2002:918258 HCAPLUS

DN 138:14206

TI Polymerizable boric acid ester compounds and their manufacture and use as polymer polyelectrolytes for electric devices

IN Yokoyama, Akihito; Yabe, Takeshi

PA NOF Corporation, Japan

SO Jpn. Kokai Tokkyo Koho, 13 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2002348323	A2	20021204	JP 2001-153804	20010523 <--
PRAI JP 2001-153804		20010523 <--		

AB The compds. useful for secondary batteries and capacitors, are the esters

of a polyoxyalkylene (meth)acrylate macromer with boric acid or its anhydride and have residual Cl content of <100 ppm. Thus, ethoxylating 2-hydroxyethyl methacrylate with ethylene oxide using BF₃-di-Et ether complex gave a macromer 284 g of which was heated with 11.6 g boric anhydride at 80° in the presence of dry air for 12 h to give a macromer borate ester (I) with Cl content <1 ppm. Polymerizing 4.00 g the I in the presence of 2.29 g LiTFSI gave a polymer electrolyte with conductivity 2.62x10⁻³ and 1.33x10⁻¹ S/m at 25 and 80°, resp.

IT 477762-06-4P

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(electrolytes; manufacture of macromer borate esters for polymer electrolytes for elec. devices)

RN 477762-06-4 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α-(2-methyl-1-oxo-2-propenyl)-ω-hydroxy-, ester with boric acid (H₃BO₃), homopolymer. (9CI) (CA INDEX NAME)

CM 1

CRN 340814-66-6

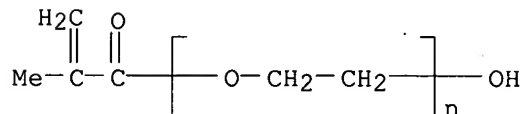
CMF (C₂ H₄ O)_n C₄ H₆ O₂ . x B H₃ O₃

CM 2

CRN 25736-86-1

CMF (C₂ H₄ O)_n C₄ H₆ O₂

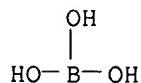
CCI PMS



CM 3

CRN 10043-35-3

CMF B H₃ O₃



IT 340814-66-6P 340814-67-7P 477762-05-3P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
(macromer; manufacture of macromer borate esters for polymer electrolytes for elec. devices)

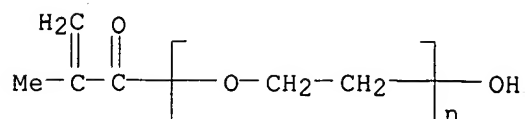
RN 340814-66-6 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α-(2-methyl-1-oxo-2-propenyl)-ω-hydroxy-, ester with boric acid (H₃BO₃) (9CI) (CA INDEX NAME)

CM 1

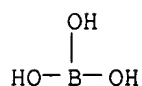
CRN 25736-86-1

CMF (C2 H4 O)_n C4 H6 O2
CCI PMS



CM 2

CRN 10043-35-3
CMF B H3 O3

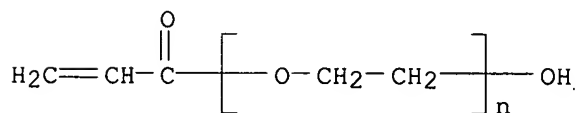


RN 340814-67-7 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α-(1-oxo-2-propenyl)-ω-hydroxy-,
ester with boric acid (H3BO3) (9CI) (CA INDEX NAME)

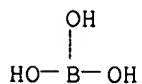
CM 1

CRN 26403-58-7
CMF (C2 H4 O)_n C3 H4 O2
CCI PMS



CM 2

CRN 10043-35-3
CMF B H3 O3

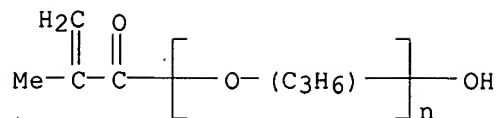


RN 477762-05-3 HCAPLUS

CN Poly[oxy(methyl-1,2-ethanediyl)], α-(2-methyl-1-oxo-2-propenyl)-
ω-hydroxy-, ester with boric acid (H3BO3) (9CI) (CA INDEX NAME)

CM 1

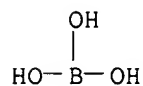
CRN 39420-45-6
CMF (C3 H6 O)_n C4 H6 O2
CCI IDS, PMS



CM 2

CRN 10043-35-3

CMF B H3 O3



IT 477594-01-7P 477762-07-5P 477762-08-6P

477762-49-5P

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
 (manufacture of macromer borate esters for polymer electrolytes for elec. devices)

RN 477594-01-7 HCAPLUS

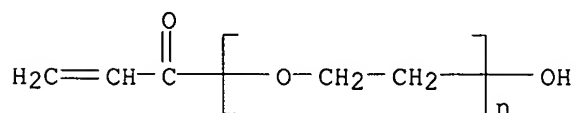
CN Poly(oxy-1,2-ethanediyl), α -(1-oxo-2-propenyl)- ω -hydroxy-,
 polymer with α -(1-oxo-2-propenyl)- ω -hydroxypoly(oxy-1,2-ethanediyl) ester with boric acid (H3BO3) (9CI) (CA INDEX NAME)

CM 1

CRN 26403-58-7

CMF (C2 H4 O)_n C3 H4 O2

CCI PMS



CM 2

CRN 340814-67-7

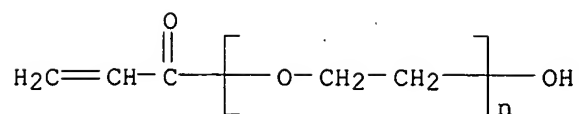
CMF (C2 H4 O)_n C3 H4 O2 . x B H3 O3

CM 3

CRN 26403-58-7

CMF (C2 H4 O)_n C3 H4 O2

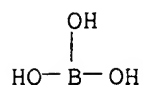
CCI PMS



CM 4

CRN 10043-35-3

CMF B H3 O3



RN 477762-07-5 HCAPLUS

CN Poly[oxy(methyl-1,2-ethanediyl)], α -(2-methyl-1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H3BO3), homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 477762-05-3

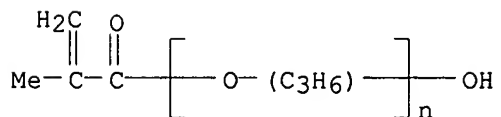
CMF (C3 H6 O)_n C4 H6 O2 . x B H3 O3

CM 2

CRN 39420-45-6

CMF (C3 H6 O)_n C4 H6 O2

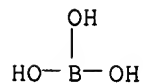
CCI IDS, PMS



CM 3

CRN 10043-35-3

CMF B H3 O3



RN 477762-08-6 HCAPLUS

CN Poly[oxy-1,2-ethanediyl], α -(1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H3BO3), homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 340814-67-7

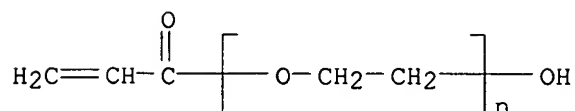
CMF (C2 H4 O)_n C3 H4 O2 . x B H3 O3

CM 2

CRN 26403-58-7

CMF (C2 H4 O)_n C3 H4 O2

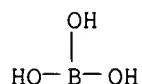
CCI PMS



CM 3

CRN 10043-35-3

CMF B H3 O3



RN 477762-49-5 HCAPLUS

CN Poly[oxy(methyl-1,2-ethanediyl)], α-(2-methyl-1-oxo-2-propenyl)-ω-hydroxy-, ester with boric acid (H3BO3), polymer with α-(1-oxo-2-propenyl)-ω-hydroxypoly(oxy-1,2-ethanediyl) ester with boric acid (H3BO3) (9CI) (CA INDEX NAME)

CM 1

CRN 477762-05-3

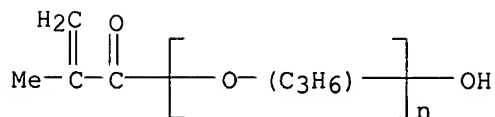
CMF (C3 H6 O)_n C4 H6 O2 . x B H3 O3

CM 2

CRN 39420-45-6

CMF (C3 H6 O)_n C4 H6 O2

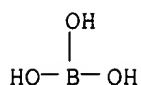
CCI IDS, PMS



CM 3

CRN 10043-35-3

CMF B H3 O3



CM 4

CRN 340814-67-7

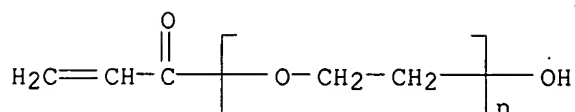
CMF (C2 H4 O)_n C3 H4 O2 . x B H3 O3

CM 5

CRN 26403-58-7

CMF (C2 H4 O)_n C3 H4 O2

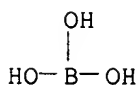
CCI PMS



CM 6

CRN 10043-35-3

CMF B H3 O3



L38 ANSWER 4 OF 10 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2002:407259 HCAPLUS

DN 137:8609

TI Secondary battery electrolyte and the battery

IN Yokoyama, Akihito; Wakiyama, Masataka

PA NOF Corporation, Japan

SO Jpn. Kokai Tokkyo Koho, 12 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2002158039	A2	20020531	JP 2000-354499	20001121 <--
PRAI	JP 2000-354499		20001121	<--	

AB The electrolyte contains an ionic compound and an organic polymer Z1[(A1O)lR1]a [Z1 = residue of a compound having 1-4 OH groups; A1 = (different) C2-4 oxyalkylene groups; l = 0-150; a = 1-4; l+a = 0-300; R1 = H, cyanoethyl group, or R3CH:CR3CO-; and R2 and R3 = H or Me] or borate ester of the polymer.

IT 340814-65-5 340814-66-6

RL: DEV (Device component use); USES (Uses)

(comps. of oxyalkylene polymers for electrolytes in secondary lithium batteries)

RN 340814-65-5 HCAPLUS

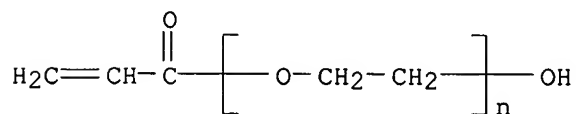
CN Poly(oxy-1,2-ethanediyl), α -(1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H3BO3) ester with α -methyl- ω -hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 26403-58-7

CMF (C2 H4 O)_n C3 H4 O2

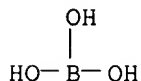
CCI PMS



CM 2

CRN 10043-35-3

CMF B H3 O3

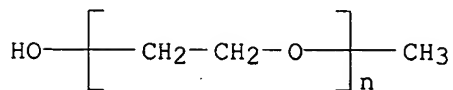


CM 3

CRN 9004-74-4

CMF (C2 H4 O)_n C H4 O

CCI PMS



RN 340814-66-6 HCAPLUS

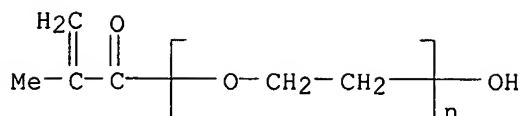
CN Poly(oxy-1,2-ethanediyl), α -(2-methyl-1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H3BO3) (9CI) (CA INDEX NAME)

CM 1

CRN 25736-86-1

CMF (C2 H4 O)_n C4 H6 O2

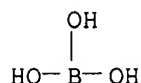
CCI PMS



CM 2

CRN 10043-35-3

CMF B H3 O3



L38 ANSWER 5 OF 10 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2001:397251 HCAPLUS

DN 135:7801

TI Secondary battery electrolytes and the batteries

IN Yokoyama, Shoichi; Wakiyama, Masataka; Kobayashi, Takao; Suwa, Kentaro

PA Nof Corporation, Japan

SO PCT Int. Appl., 53 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2001039316	A1	20010531	WO 2000-JP8254	20001122 <--
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CZ, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
	RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG				
	JP 2001155771	A2	20010608	JP 1999-332586	19991124 <--
	JP 2001273925	A2	20011005	JP 2000-87754	20000328 <--
	CA 2392543	AA	20010531	CA 2000-2392543	20001122 <--
	AU 2001015495	A5	20010604	AU 2001-15495	20001122 <--
	EP 1258938	A1	20021120	EP 2000-977877	20001122 <--
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR				
	US 6833220	B1	20041221	US 2002-130952	20020524 <--
PRAI	JP 1999-332586	A	19991124	<--	
	JP 2000-87754	A	20000328	<--	
	WO 2000-JP8254	W	20001122	<--	

AB The electrolytes contain a n ionic compound and a polymer, where the polymer is Z1[(A1O)lR1]a (R1 = cyanoethyl, C1-12 hydrocarbon group, or H; Z1 = a residue of a compound having 1-6 OH groups; A1O is ≥1 C2-4 oxyalkylene group; l = 0-600, a = 1-6, and a+l = 0-600), or its borate ester or Z2[(A2O)mR2]b (R2 = H, cyanoethyl or R3CH:CR4CO; Z2 = OH

or residue of a compound having 1-4 OH groups; A20 is ≥ 1 C2-4 oxyalkylene group; R3 and R4 = H or Me; m = 0-150, b = 1-4, and m+b = 0-300).

IT 340814-65-5 340814-66-6 340814-67-7

RL: DEV (Device component use); USES (Uses)

(compns. of oxyalkylene polymer electrolytes for secondary lithium batteries)

RN 340814-65-5 HCAPLUS

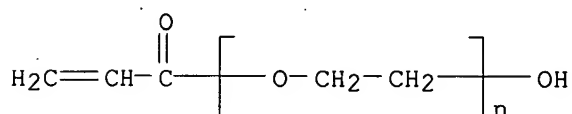
CN Poly(oxy-1,2-ethanediyl), α -(1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H3BO3) ester with α -methyl- ω -hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 26403-58-7

CMF (C2 H4 O)_n C3 H4 O2

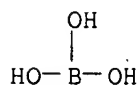
CCI PMS



CM 2

CRN 10043-35-3

CMF B H3 O3

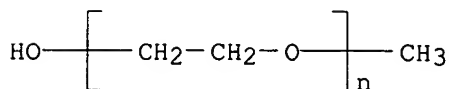


CM 3

CRN 9004-74-4

CMF (C2 H4 O)_n C H4 O

CCI PMS



RN 340814-66-6 HCAPLUS

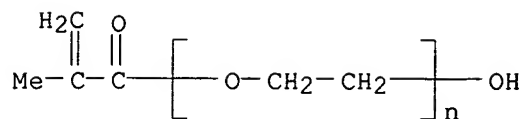
CN Poly(oxy-1,2-ethanediyl), α -(2-methyl-1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H3BO3) (9CI) (CA INDEX NAME)

CM 1

CRN 25736-86-1

CMF (C2 H4 O)_n C4 H6 O2

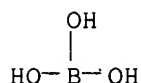
CCI PMS



CM 2

CRN 10043-35-3

CMF B H3 O3



RN 340814-67-7 HCAPLUS

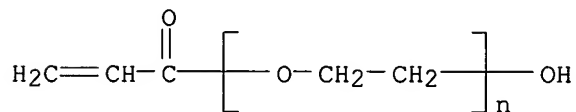
CN Poly(oxy-1,2-ethanediyl), α -(1-oxo-2-propenyl)- ω -hydroxy-, ester with boric acid (H3BO3) (9CI) (CA INDEX NAME)

CM 1

CRN 26403-58-7

CMF (C2 H4 O)_n C3 H4 O2

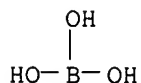
CCI PMS



CM 2

CRN 10043-35-3

CMF B H3 O3



RETABLE

Referenced Author (RAU)	Year (RPY)	VOL (RVL)	PG (RPG)	Referenced Work (RWK)	Referenced File
Mitsubishi Chemical Cor	1999			JP 11219726 A	HCAPLUS
Nippon Oil Co Ltd	1992			US 5326657 A	HCAPLUS
Nippon Oil Company Ltd	1993			JP 536305 A	
Toshiba Battery Co Ltd	1993			JP 05151992 A	HCAPLUS
Ube Industries Ltd	1993			JP 05315007 A	HCAPLUS

L38 ANSWER 6 OF 10 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1987:85925 HCAPLUS

DN 106:85925
 TI Neutron-shielding transparent resin
 IN Ida, Kozo; Azegami, Kyotaka
 PA Mitsubishi Rayon Co., Ltd., Japan
 SO Jpn. Kokai Tokkyo Koho, 4 pp.
 CODEN: JKXXAF
 DT Patent
 LA Japanese
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 61215610	A2	19860925	JP 1985-56108	19850322 <--
PRAI	JP 1985-56108		19850322	<--	

AB Hydroxyalkyl or glycidyl (meth)acrylates are treated with H₃BO₃, its esters, or BO_x and polymerized to give the title resins, containing 0.2-6% B. Thus, 3 mol 2-hydroxyethyl methacrylate and 1 mol H₃BO₃ were esterified 3 h at 80° and polymerized to a polymer with neutron shielding 1200% of that of poly(Me methacrylate).

IT 106946-68-3

RL: USES (Uses)

(neutron shields, manufacture of transparent)

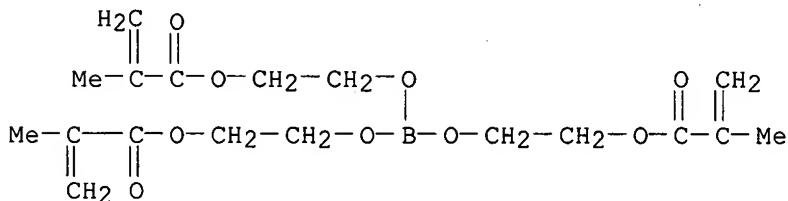
RN 106946-68-3 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, borylidynetris(oxy-2,1-ethanediyl) ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 42175-72-4

CMF C18 H27 B O9

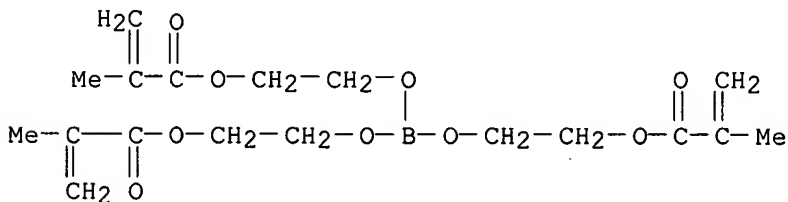


IT 42175-72-4P

RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation of)

RN 42175-72-4 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, borylidynetris(oxy-2,1-ethanediyl) ester
 (9CI) (CA INDEX NAME)



L38 ANSWER 7 OF 10 HCAPLUS COPYRIGHT 2006 ACS on STN
 AN 1981:182444 HCAPLUS

jan delaval - 30 october 2006

DN 94:182444
 TI Neutron shielding materials
 PA Kyowa Gas Chemical Industry Co., Ltd., Japan
 SO Jpn. Kokai Tokyo Koho, 8 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 55144597	A2	19801111	JP 1979-53032	19790427 <--
	JP 59017398	B4	19840420		
PRAI	JP 1979-53032	A	19790427	<--	

AB Mixts. composed of (1) ≥ 1 monomer selected from alkyl(C1-4) methacrylates and styrene and (2) boric acid ester C3-16 polyols are copolymd. to give n shielding materials. The composition is formulated such that polyol/B ratio and B content in the shield are 0.6-4.0 and 1-6 weight%, resp. Thus, H3BO3 61.8, 1,2-propanediol 152, and Me methacrylate 250 parts were heated at 60°, then excess Me methacrylate and the product H2O were removed by evaporation to give an ester solution. Then, Me methacrylate 2, ethylene glycol dimethacrylate 2 parts, and a polymerization initiator were added to the ester solution, and copolymn. was carried out to give a n shielding material having good optical transparency, good mech. strength, and good n shielding efficiency.

IT 77466-47-8

RL: PROC (Process)

(polymer composition containing, for neutron shield)

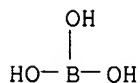
RN 77466-47-8 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-hydroxyethyl ester, homopolymer, ester with boric acid (H3BO3) (9CI) (CA INDEX NAME)

CM 1

CRN 10043-35-3

CMF B H3 O3



CM 2

CRN 25249-16-5

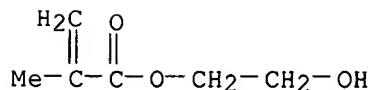
CMF (C6 H10 O3)x

CCI PMS

CM 3

CRN 868-77-9

CMF C6 H10 O3



L38 ANSWER 8 OF 10 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1974:49101 HCAPLUS

DN 80:49101

TI Polymerizing acrylonitrile

IN Yoshino, Tsuneo; Kenjo, Hideki

PA Toray Industries, Inc.

SO Jpn. Kokai Tokkyo Koho, 3 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 48060183	A2	19730823	JP 1971-95942	19711130 <--
	JP 55001286	B4	19800112		
PRAI	JP 1971-95942	A	19711130	<--	

AB Borates with polymerizable ester groups were used as temporary crosslinking agents in an acrylonitrile(I) continuous polymerization procedure in

which polar solvents were used to dissolve the polyacrylonitrile(II). Thus, I 100, Me2SO 25, tris(2-methacryloyloxyethyl) borate [42175-72-4] 3, azobis(2,4-dimethylvaleronitrile) 0.6 and dodecyl mercaptan 0.2 part were placed into a PVC tubular cell which was inserted between steel plates and kept 10 min at 70, 80, and 90.deg., and the product stretched 2000% in 4:6 DMF-H2O at 95.deg. to give a transparent filament.

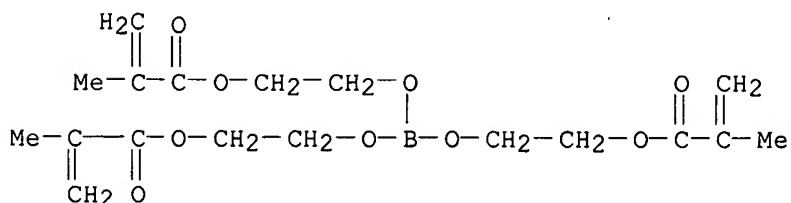
IT 42175-72-4

RL: MOA (Modifier or additive use); USES (Uses)

(crosslinking agents, temporary, in acrylonitrile polymer manufacture in polar solvents)

RN 42175-72-4 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, borylidynetris(oxy-2,1-ethanediyl) ester (9CI) (CA INDEX NAME)



L38 ANSWER 9 OF 10 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1973:467429 HCAPLUS

DN 79:67429

TI Unsaturated polyester resins which cure in the presence of water and which contain at least one dehydro monomer

IN Koch, Stanley D.; Gerber, Arthur H.

PA Horizons Inc.

SO U.S., 4 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI US 3743686 A 19730703 US 1971-154657 19710618 <--
 PRAI US 1971-154657 A 19710618 <--

AB Addition of hydrolyzable monomers, e.g. unsatd. anhydrides, mixed anhydrides of unsatd. carboxylic and boron or silicon acids, or borate esters, to unsatd. polyesters containing styrene or Me methacrylate gave resin compns. which could be cured at room temperature in the presence of H₂O. Thus, a mixture

of H₃BO₃ 6.18, acrylic anhydride 40.32, anhydrous ZnCl₂ 0.004, and hydroquinone 0.020 g was heated at 20-70.deg. to give acrylic acid-orthoboric acid mixed trianhydride (I) [41621-92-5]. Addition of 13 g I to a mixture of maleic anhydride-phthalic anhydride-propylene glycol copolymer [25037-66-5] 75, styrene 12, and H₂O 3 g gave a resin with SPI gel time of 6.5 min compared with >1 hr for a similar polyester composition not containing I.

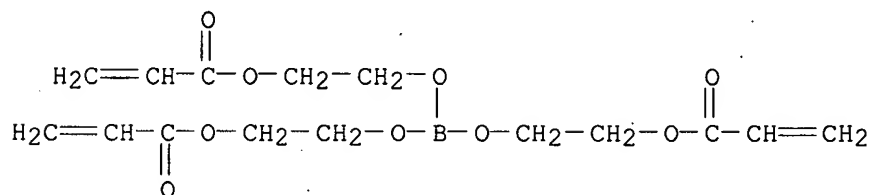
IT 32577-32-5 42175-71-3 42175-72-4

RL: MOA (Modifier or additive use); USES (Uses)

(crosslinking agents, for unsatd. polyesters in presence of water)

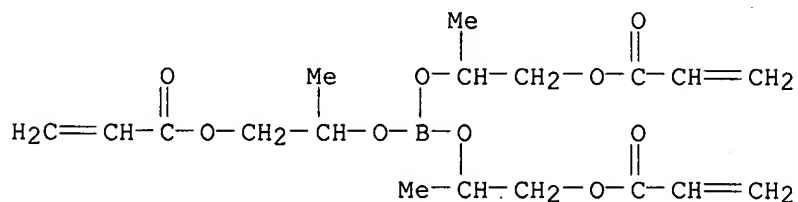
RN 32577-32-5 HCAPLUS

CN 2-Propenoic acid, borylidynetris(oxy-2,1-ethanediyl) ester (9CI) (CA INDEX NAME)



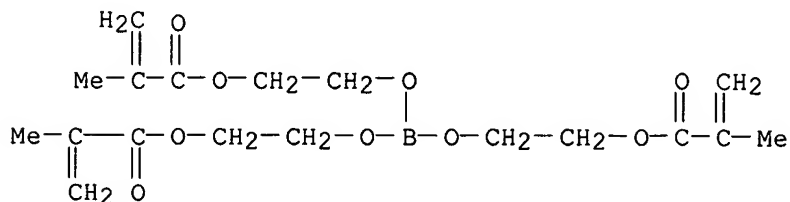
RN 42175-71-3 HCAPLUS

CN 2-Propenoic acid, borylidynetris[oxy(2-methyl-2,1-ethanediyl)] ester (9CI).
 (CA INDEX NAME)



RN 42175-72-4 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, borylidynetris(oxy-2,1-ethanediyl) ester
 (9CI) (CA INDEX NAME)



L38 ANSWER 10 OF 10 HCAPLUS COPYRIGHT 2006 ACS on STN
 AN 1971:126877 HCAPLUS
 DN 74:126877

TI Vinylthioethyl borates for flameproofing polyesters
 IN Naarmann, Herbert; Hartmann, Heinrich
 PA Badische Anilin- & Soda-Fabrik AG
 SO Ger. Offen., 6 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	DE 1901058	A	19710211	DE 1969-1901058	19690110 <--
	DE 1901058	B2	19731018		
	DE 1901058	C3	19740522		
	FR 2032310	A5	19701127	FR 1970-713	19700109 <--
PRAI	DE 1969-1901058	A	19690110	<--	

GI For diagram(s), see printed CA Issue.

AB The borate flameproofing agents, (H₂C:CHSCH₂CH₂O)_nB(OCH₂CH₂SH)_{3-n}, where n is 2 or 3, effectively reduce the flammability of polyesters. Borates used include tris(2-vinylthioethyl)borate (I) and bis(2-vinylthioethyl)-2-mercaptoethyl borate (II). A polyester was prepared from phthalic acid 1, maleic acid 2, and propylene glycol 3 parts by heating at 180° to 50 acid number, dilution with styrene to 66% polyester and acid number 30, addition of

0.4% cyclohexanone peroxide and 0.04% Co naphthenate, and hardening. A mixture of this polyester with 1.8% II ignited in 20-5 sec. (ASTM DG 35-56T) with 15 sec smoldering time as compared with 15-20 and .apprx.20 sec, resp., for a polyester containing 1.8% H₃BO₃. The borates are prepared by treating B(OH)₃ with H₂C:CHS(CH₂)₂OH and HS(CH₂)₂OH.

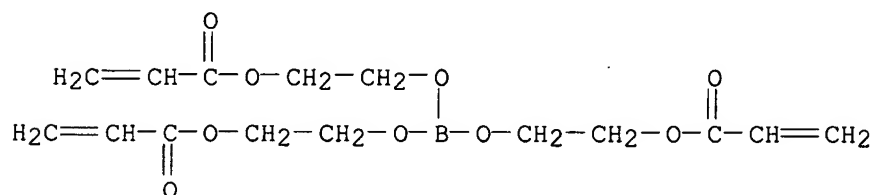
IT 32577-32-5

RL: USES (Uses)

(fireproofing with, of polyesters)

RN 32577-32-5 HCAPLUS

CN 2-Propenoic acid, borylidynetris(oxy-2,1-ethanediyl) ester (9CI) (CA INDEX NAME)



=> d 139 bib abs hitstr retable tot

L39 ANSWER 1 OF 3 HCAPLUS COPYRIGHT 2006 ACS on STN
 AN 2006:1010866 HCAPLUS
 DN 145:380336

TI Gel electrolytes of borate acrylate polymers, and nonaqueous electrolyte secondary batteries using them

IN Okumura, Takefumi; Nishimura, Shin; Iwayasu, Norio; Kono, Kazushige; Yokoyama, Akihito; Mizutani, Masato; Ito, Tetsuya

PA Hitachi Ltd., Japan; NOF Corporation

SO Jpn. Kokai Tokkyo Koho, 21pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2006261024	A2	20060928	JP 2005-79288	20050318
PRAI	JP 2005-79288		20050318		

AB The gel electrolytes have matrix polymers manufactured by polymerizing polymerizable

group-terminated borate esters $Z1(AO)lOB[O(AO)mZ2]O(AO)nZ3$ ($Z1-Z3$ = polymerizable functional group, C1-10 hydrocarbyl; average molar content of the C1-10 hydrocarbyl in $Z1-Z3$ = 1.0-2.5 mol; AO = C2-4 oxyalkylene; l, m, n = 0-100; $l + m + n = 1-300$). Alternatively, the gel electrolytes further contain $X1(AO)\alpha OB[O(AO)\beta X2]O(AO)\gamma X3$ ($X1-X3$ = C1-10 hydrocarbyl; AO = same as above; $\alpha + \beta + \gamma = 1-300$). Secondary batteries with good charging properties are provided with this invention.

IT 866555-99-9P

RL: DEV (Device component use); IMF (Industrial manufacture); POF (Polymer in formulation); PREP (Preparation); USES (Uses)
(gel electrolytes of borate acrylate polymers for nonaq. electrolyte secondary batteries)

RN 866555-99-9 HCAPLUS

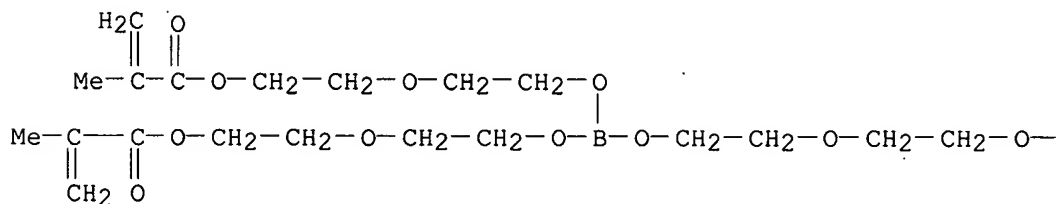
CN 2-Propenoic acid, 2-methyl-, borylidynetris(oxy-2,1-ethanediyl) ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

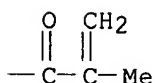
CRN 866555-98-8

CMF C24 H39 B 012

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PAGE 1-B



L39 ANSWER 2 OF 3 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2006:77298 HCAPLUS

DN 144:153448

TI Electrode for secondary polymer electrolyte battery and the battery
 IN Okumura, Takefumi; Nishimura, Shin; Iwayasu,
 Norio; Yokoyama, Shoichi; Itoh, Tetsuya; Yabe,
 Takeshi; Ichimiya, Kengo
 PA Hitachi, Ltd., Japan; NOF Corporation
 SO PCT Int. Appl., 44 pp.
 CODEN: PIXXD2
 DT Patent
 LA Japanese
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2006009284	A1	20060126	WO 2005-JP13671	20050720
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				

PRAI JP 2004-211412 A 20040720

AB The battery has a cathode containing a cation-intercalating cathode active mass, an anode containing a cation-intercalating anode active mass, and an electrolyte layer interposed between the cathode and the anode and composed of an ion-conductive polymer for transferring the cations; where the cathode and/or the anode comprises a B-cong. organic compound as a binder component; and the cathode and/or anode active mass is treated with silane, Al, or Ti for facilitating intercalation/decalation of cations, thereby suppressing decrease in charge/discharge capacity.

IT 866555-98-8

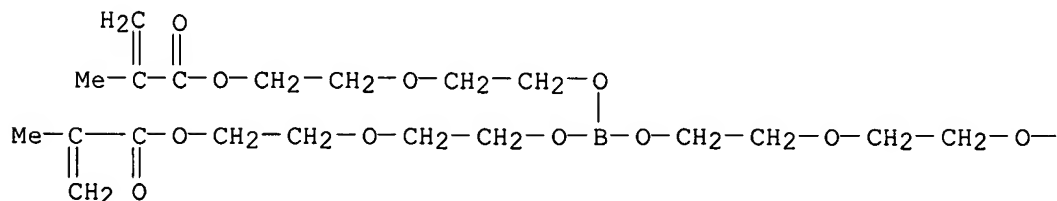
RL: DEV (Device component use); USES (Uses)

(electrodes having boron-containing organic compound binders and modified active mass for secondary lithium batteries)

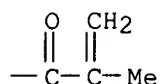
RN 866555-98-8 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, borylidynetris(oxy-2,1-ethanediyl) ester (9CI) (CA INDEX NAME)

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RETABLE

Referenced Author (RAU)	Year (RPY)	VOL (RVL)	PG (RPG)	Referenced Work (RWK)	Referenced File
Hitachi Ltd	2004			US 2004101758 A	
Hitachi Ltd	2004			JP 2004186150 A	HCAPLUS
Hitachi Ltd	2004			FR 2847721 A1	HCAPLUS
Hitachi Ltd	2005			JP 2005285416 A	HCAPLUS
Kabushiki Kaisha Samsun	1999			JP 11-329435 A	HCAPLUS
Kabushiki Kaisha Samsun	1999			US 6218050 B1	HCAPLUS
Sony Corp	1998			JP 10-125307 A	HCAPLUS
Sumitomo Chemical Co Lt	1996			JP 08-111243 A	HCAPLUS
Sumitomo Chemical Co Lt	1996			US 5571638 A	HCAPLUS
Sumitomo Chemical Co Lt	1996			EP 652602 A2	HCAPLUS
Toyota Motor Corp	2004			JP 20046237 A	
Yuasa Corp	2003			JP 200392138 A	

L39 ANSWER 3 OF 3 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2005:1106331 HCAPLUS

DN 143:389780

TI Secondary batteries with high discharge capacity and cycle efficiency, and cathode and anodes therefor

IN Okumura, Takefumi; Nishimura, Shin; Iwayasu, Norio; Yokoyama, Akihito; Ito, Tetsuya; Yabe, Takeshi; Ichinomiya, Kengo

PA Hitachi Ltd., Japan; NOF Corporation

SO Jpn. Kokai Tokkyo Koho, 20 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2005285416	A2	20051013	JP 2004-94798	20040329
PRAI	JP 2004-94798		20040329		

AB The batteries contain ion-conductive polymer electrolyte layers and B-containing organic compds. as binders in cathodes and/or anodes.

IT 866555-99-9DP, lithium complex

RL: DEV (Device component use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)

(secondary batteries containing B-containing organic compds. as binders in cathodes and/or anodes)

RN 866555-99-9 HCAPLUS

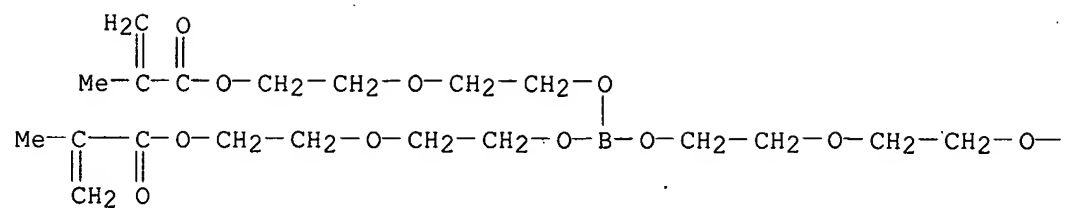
CN 2-Propenoic acid, 2-methyl-, borylidynetris(oxy-2,1-ethanediyl) ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

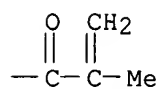
CRN 866555-98-8

CMF C24 H39 B 012

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